

Name: _____

Percent Composition

1. Find the % composition of H_2O
2. Find the % composition of FeSO_4
3. Find the % composition of $\text{C}_2\text{H}_5\text{OH}$
4. Find the % composition of $(\text{NH}_4)_3\text{PO}_4$
5. Find the % composition of water in $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$
6. Analysis of a compound shows that it consists of 352g of Cu and 48g of oxygen. What is the % composition of the compound?
7. If you had 126.9g of H_2CO_3 , what mass would be due to hydrogen? What mass would be due to carbon? Oxygen?